

## Mark schemes

## Q1.

- (a) colourless

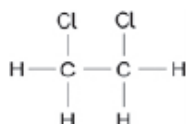
*ignore clear*

1

- (b) damp litmus paper

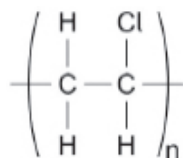
1

- (c)



1

- (d)



1

- (e)
- $1.5 \times 10$
- $0.9 \times 10$

1

(ratio =) 15 : 9

1

= 5 : 3

*allow correct determination of the simplest whole number ratio from an attempt at a density ratio*

1

**alternative approach**

(ratio =)

$$\frac{1.5}{0.9} \quad \frac{0.9}{0.9} \quad (1)$$

1.666 : 1 (1)

= 5 : 3 (1)

*allow correct determination of the simplest whole number ratio from an attempt at a density ratio*

- (f) the pipes will melt  
**or**  
 the polymers will melt

*allow the melting point of both polymers is below 300°C*

1

- (g) oil is non-renewable  
**or**  
 paper is obtained from a renewable source  
*allow oil is finite*

1

**[9]****Q2.**

- (a) alloy

1

reinforcement

1

- (b) burning (of methane) releases carbon dioxide  
*allow burning methane*  
*ignore methane is a greenhouse gas*

1

decomposition (of limestone) releases carbon dioxide  
*allow decomposition of limestone*

1

- (c) hydrochloric acid

1

limewater

1

- (d) (pre-stressed concrete) can bear the weight of (heavy) traffic  
*allow converse for plain concrete*  
*allow (pre-stressed concrete) bridge is less likely to collapse*

1

(because pre-stressed concrete is) stronger  
*do **not** accept (because prestressed concrete is) more dense*

1

- (e) any **two** from:  
 (plain concrete slabs)  
 • are cheaper  
 • will be lighter (to transport / lay)  
 • do not need to carry vehicles  
*allow converse for pre-stressed concrete*

2

**[10]**

**Q3.**

- (a) (the poly(propene) beaker will begin to) melt  
*allow poly(propene) has a low melting point*

1

(the poly(propene) beaker will) burn / ignite  
*allow poly(propene) is flammable*

1

- (b) (poly(propene) beakers are) less easily broken  
*allow (poly(propene) beakers are) less likely to shatter*

*allow (poly(propene) beakers are) tougher*

*allow (poly(propene) beakers have a) higher resistance to impact*

1

- (c) boron trioxide

1

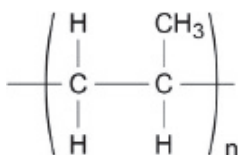
- (d)

Symbol for element	Name of element	Number of atoms of element in one molecule of propene
C	carbon	3
H	hydrogen	6

*if no other mark awarded allow 1 mark for a correct column*

2

- (e)



1

- (f) (**Stage 1** is) fractional distillation

1

(**Stage 2** is) cracking

1

(**Stage 3** is) polymerisation

1

- (g) alkene

1

monomer

1

**[12]**

**Q4.**

(a) carbon 1

(b) chromium 1

nickel 1

(c) hard 1

resistant to corrosion 1

*in either order*

(d) (percentage of titanium =  $100 - 3.0 - 2.5 = 94.5$  (%)) 1

$$(\text{mass}) = \frac{94.5}{100} \times 5.0$$

*allow correct use of incorrectly determined percentage of titanium*

$$= 4.725 \text{ (kg)}$$

*allow 4.7 / 4.73 (kg)*

**alternative approach:**

$$(\text{mass of Al} + \text{V} = \frac{5.5}{100} \times 5.0 =)$$

$$(\text{mass of titanium} =) 5 - 0.275 \text{ (1)}$$

*allow correct use of incorrectly determined mass of Al and V*

$$= 4.725 \text{ (kg) (1)}$$

*allow 4.7 / 4.73 (kg)*

(e) (both are) strong 1  
*allow (both contain) more aluminium*

(f) tin is toxic 1  
*allow tin reacts in the body*

**[10]**